

Decentralized CO₂- neutral Bio-Methanol production system



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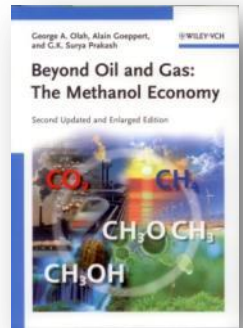
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Renewable energy challenges



- New and efficient ways to store energy needed
- A new, viable alternative approach is based on **methanol**
 - Suggested some years ago by **Prof. Dr. George Olah**¹, Distinguished Professor and Director of the Loker Hydrocarbon Research Institute at University of Southern California.
 - Awarded Nobel Prize in Chemistry in 1994
- “Methanol is the simplest, safest and easiest to **store** and **transport** liquid oxygenated hydrocarbon”



¹see book: Beyond Oil and Gas: The Methanol Economy, ISBN 978-3-527-32422-4



Methanol (CH₃OH)

hydrogen **methanol** ethanol

- Currently almost exclusively prepared from synthesis gas (syn-gas, a mixture of CO and H₂)
- Obtained from incomplete combustion and reforming of fossil fuels - mainly natural gas and coal - with steam
- Methanol can also be prepared from
 - a. biomass (wood, agricultural by-products (e.g. straw), municipal waste)
 - b. fermentation (bio)- and sewage gas
 - c. environmental recycling of CO₂
- Our Methanol production efficiency is 10x of Ethanol

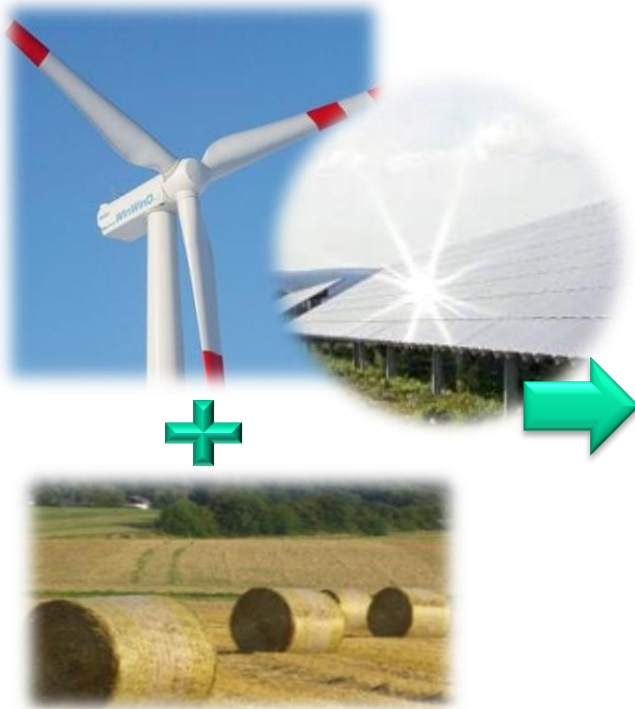


AEN Solution Concept

Decoupling consumption from generation

be local and autonomous

Source



Consumption



Bio-Methanol
Storage



Economic Green Electricity Storage



Methanol as Fuel in ICE¹

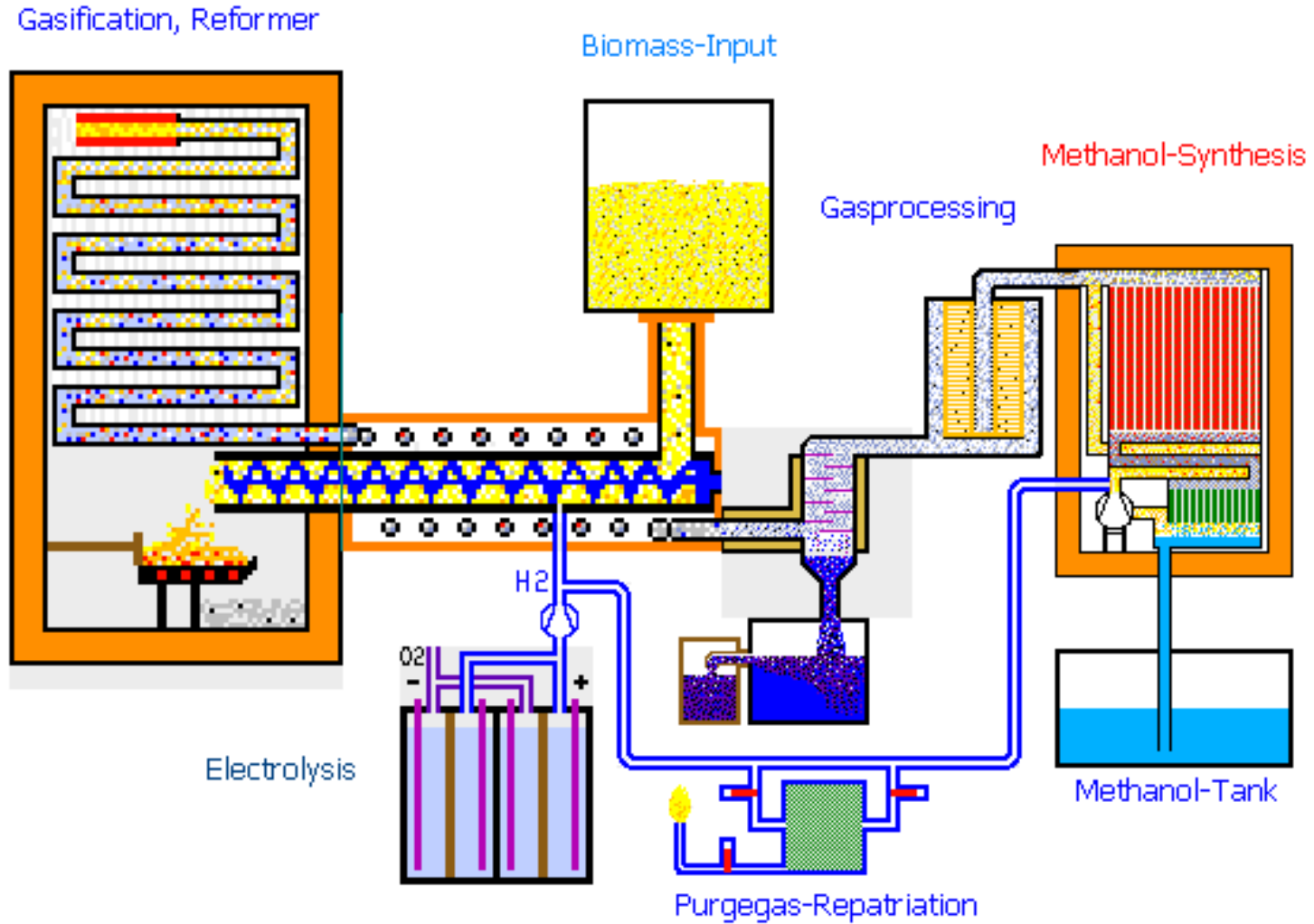
- Methanol is a simple chemical, in contrast to gasoline
 - Approx. ½ the energy density of gasoline
 - Has a higher octane rating of 100 (average of RON 107 and MON 92)
 - Fuel/air mixture can be more compressed before ignition (10-11:1 vs. 8-9:1)
 - Higher “flame speed” – faster and more complete fuel combustion
- **Important: current gasoline engines can be rel. simply modified**
- Methanol-specific engines
 - Could be more compact and lighter
 - 3.7 x higher latent heat of vaporization – less cooling
 - Low overall emissions of air pollutants (NO_x, SO₂, particulates)
 - CO₂ neutral, due to biomass and green electricity



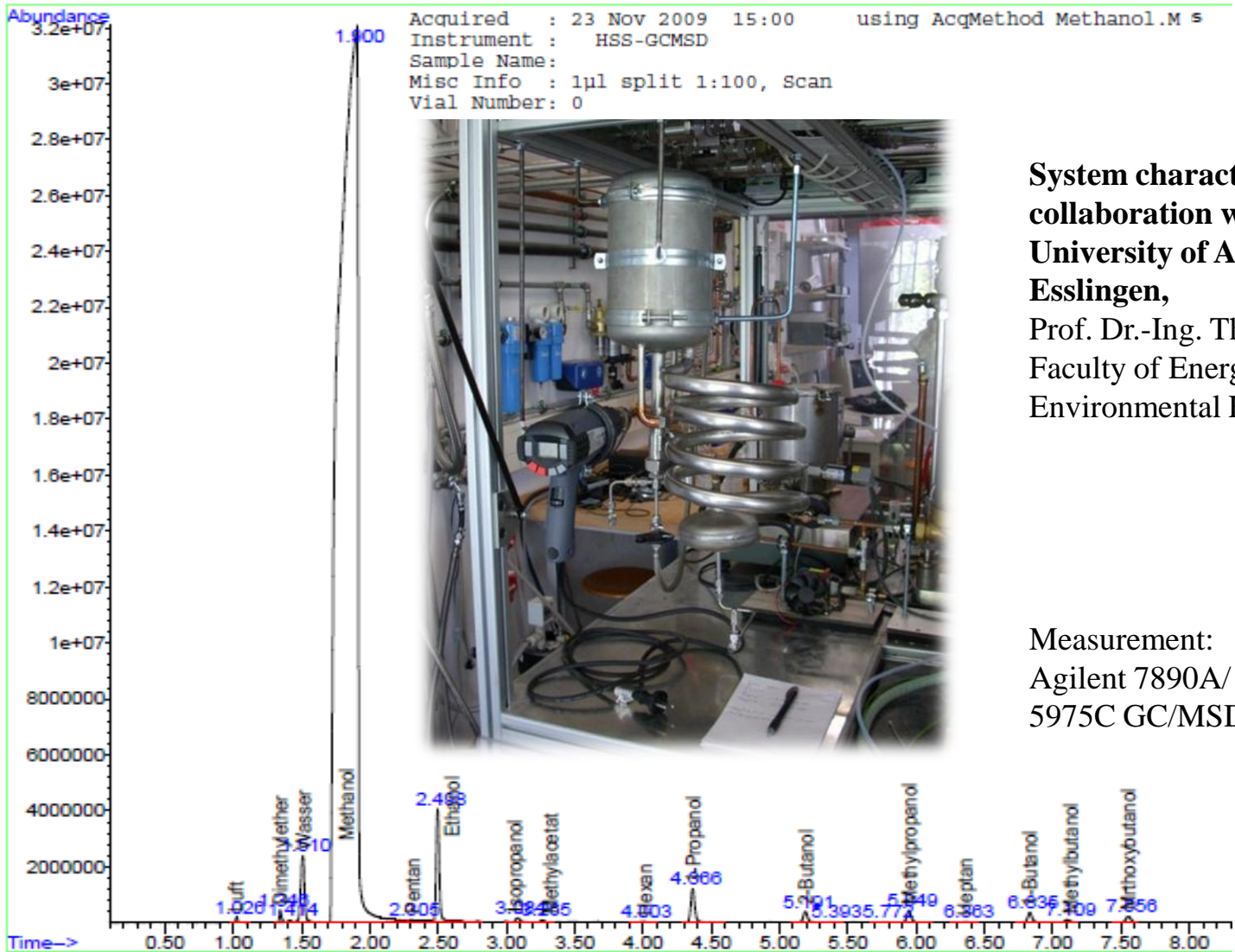
Figure 11.2 Lotus Exige 270E tri-fuel – able to run on any mixture



Electricity-Biomass-Methanol System



Prototype System



System characterization in collaboration with the University of Applied Sciences Esslingen,
 Prof. Dr.-Ing. Thomas Rohrbach
 Faculty of Energy & Environmental Engineering

Measurement:
 Agilent 7890A/
 5975C GC/MSD



Storage Efficiency

Mass balance

Biomass - steam reforming and H2-Electrolysis

kg Biomass Input	1,0	kg biomass (BM)
kg water H2O total input	1,0	kg H2O / kg BM
kg water H2O to syn-gas steam-reforming	0,25	kg H2O / kg BM
kg water H2O to H2-electrolysis using electricity	0,75	kg H2O / kg BM
kg H2 in to syn-gas from H2-electrolysis using electricity	0,083	kg H2 / kg BM
Output [kg] methanol / kg biomass	1,33	
Output [liter] methanol / kg biomass	1,69	

Energy balance

Biomass in KWh	5,0	KWh / kg BM
Required electricity for H2-electrolysis in KWh	4,2	KWh / kg BM
Required electricity in KWh / liter methanol	2,5	KWh / L MeOH
Efficiency	80%	
Methanol output /electricity input	200%	KJ MeOH / KJ e-power



In summary - we provide ...

- Bio-Methanol
 - An effective green electricity storage
 - A CO₂ neutral, price competitive transportation fuel
- Utilizes domestic biomass and biogas
- Compact, economical and scalable system

Thanks

Questions?
Comments?
Interests?

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